

Quantum Mechanics & Quantum Computation

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Lecture 1: Introduction

Double Slit Experiment



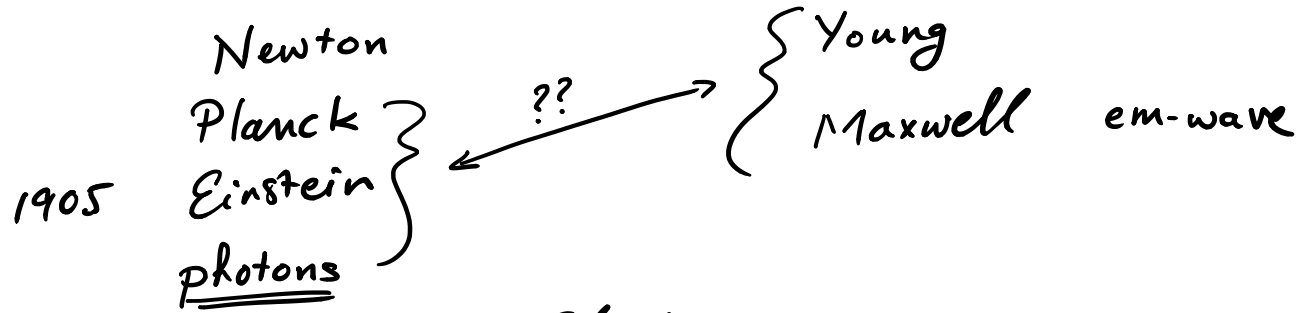
Successful

- * Conductivity of metals
- * Transparency of glass
- * Colors
- * Chemical reactions
- ⋮

Young's Double Slit Experiment.

Rain of particles/bullets

Wave



Electrons

Electron diffraction -



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$$\frac{1}{\sqrt{2}} |\text{cat}\rangle + \frac{1}{\sqrt{2}} |\text{dog}\rangle$$

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Double-slit experiment

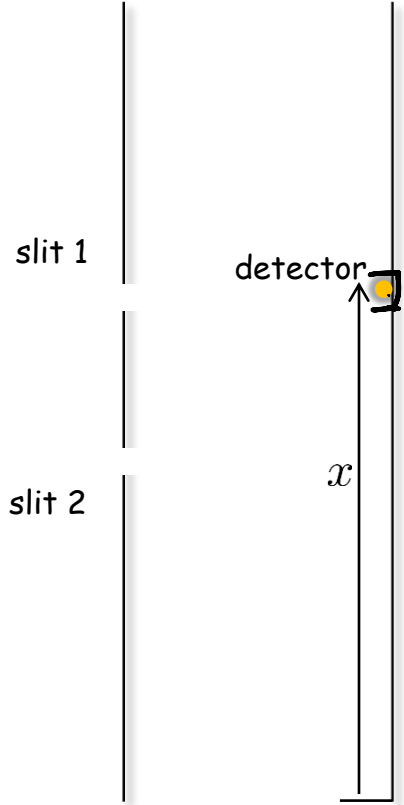
Wave or Particle?

Light: rain of particles
waves

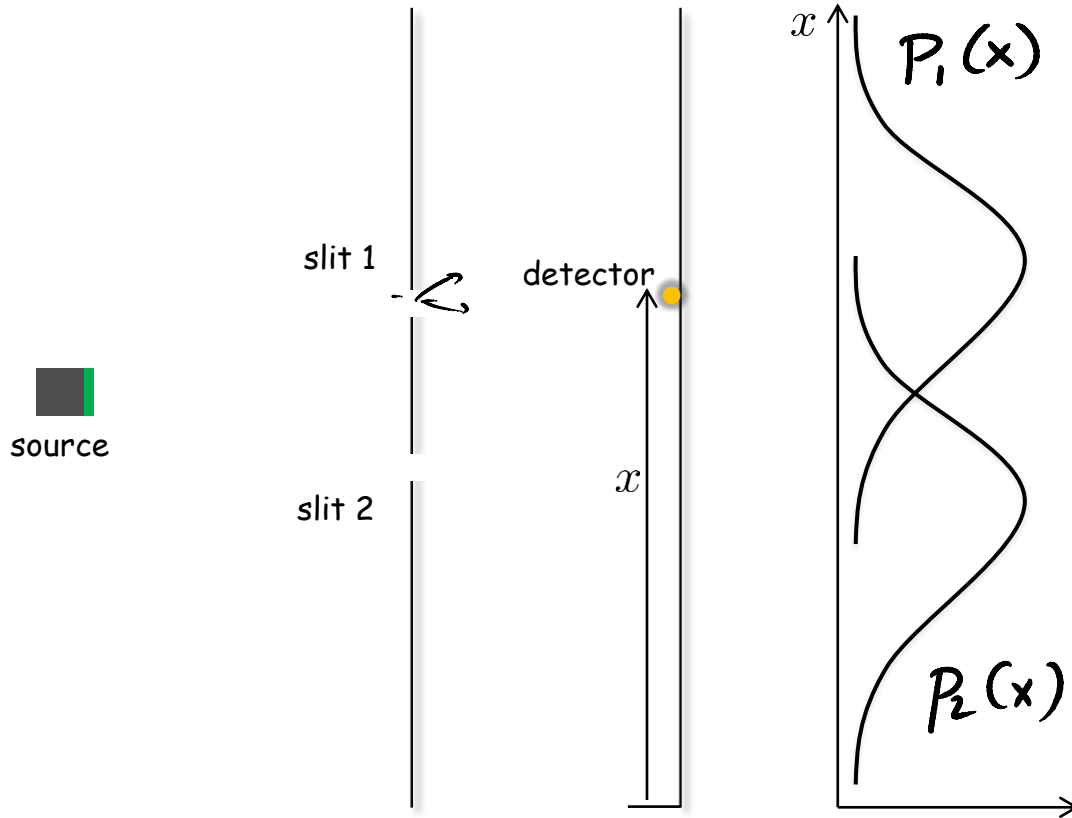
Electrons: particles
wave (electron diffraction)

Double-slit experiment

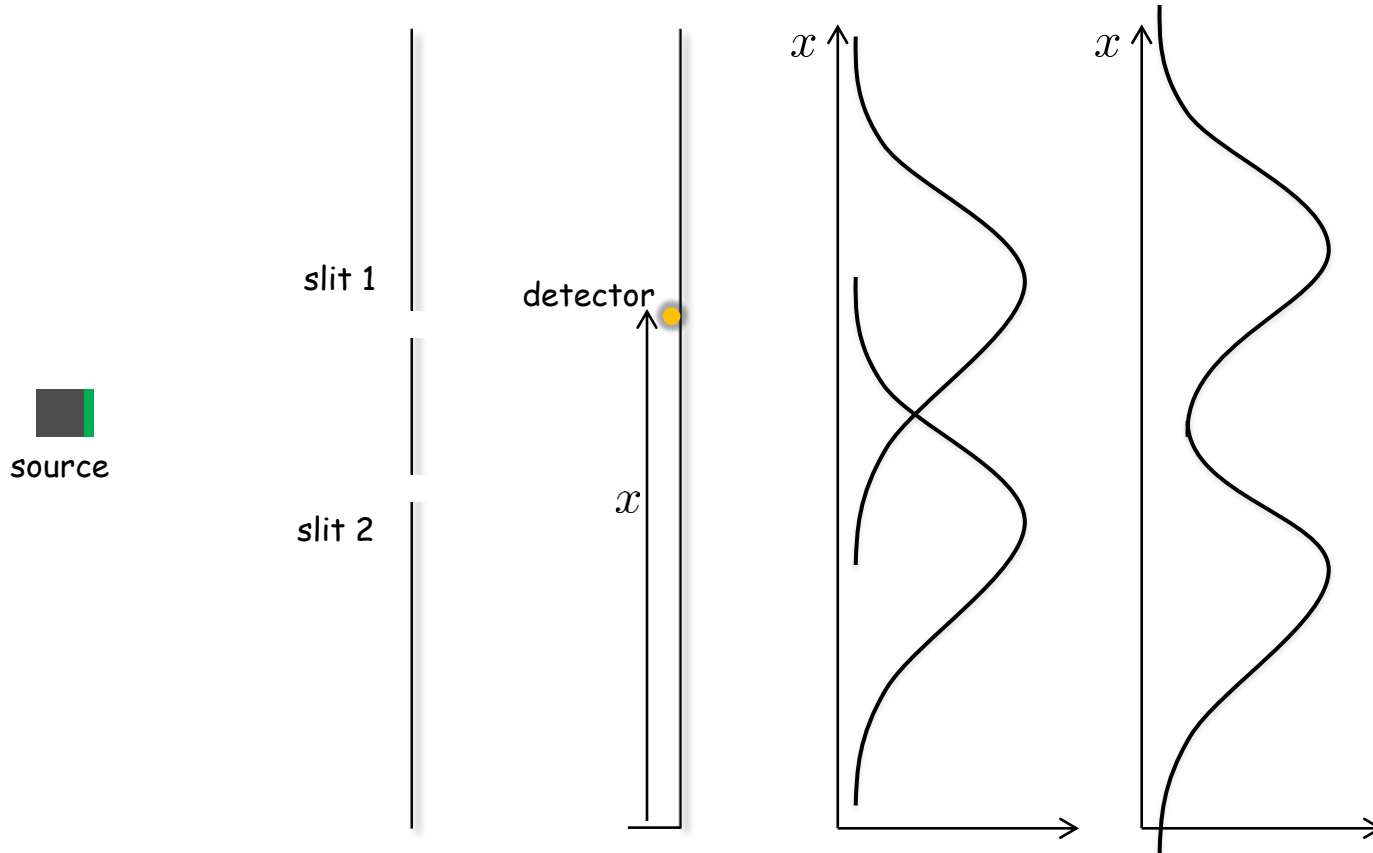
photons
electrons



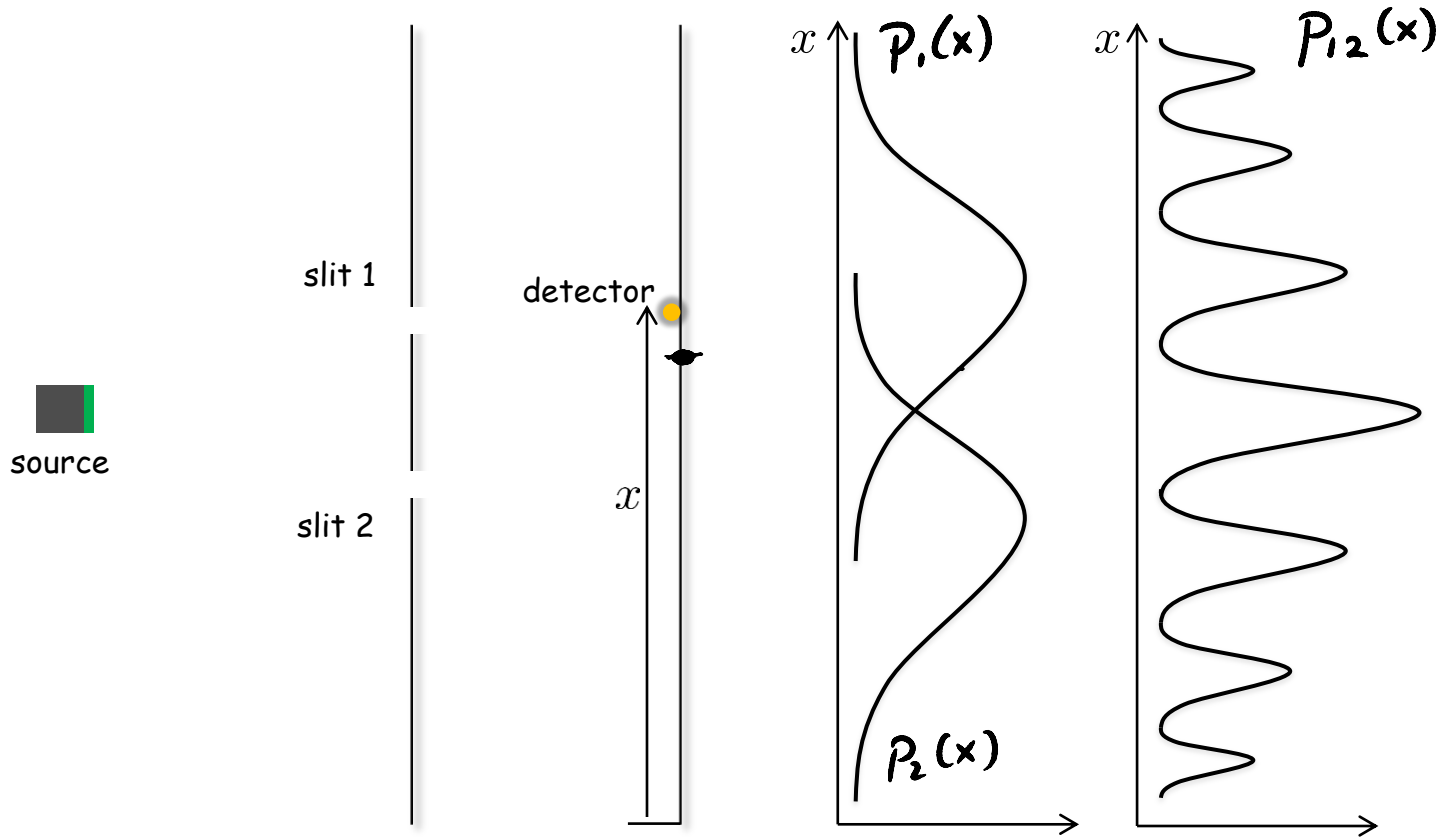
Double-slit experiment



Double-slit experiment



Double-slit experiment



I can safely say that no one understands quantum mechanics.
- Richard Feynman

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$$\frac{1}{\sqrt{2}} |\text{cat}\rangle + \frac{1}{\sqrt{2}} |\text{dog}\rangle$$

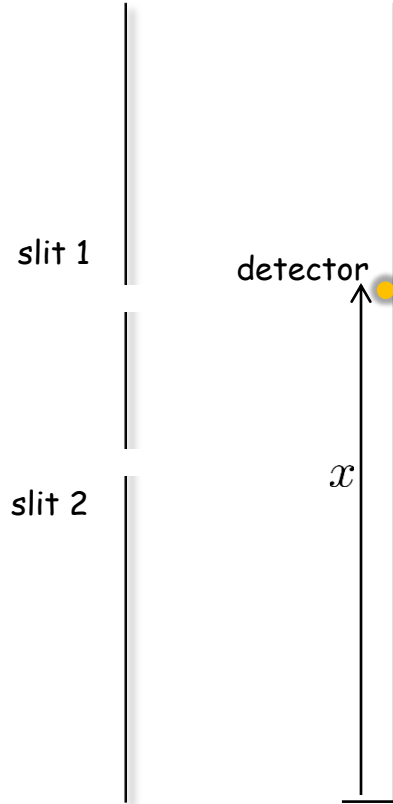
Lecture 1: Introduction

Double-slit experiment

Double-slit experiment



source

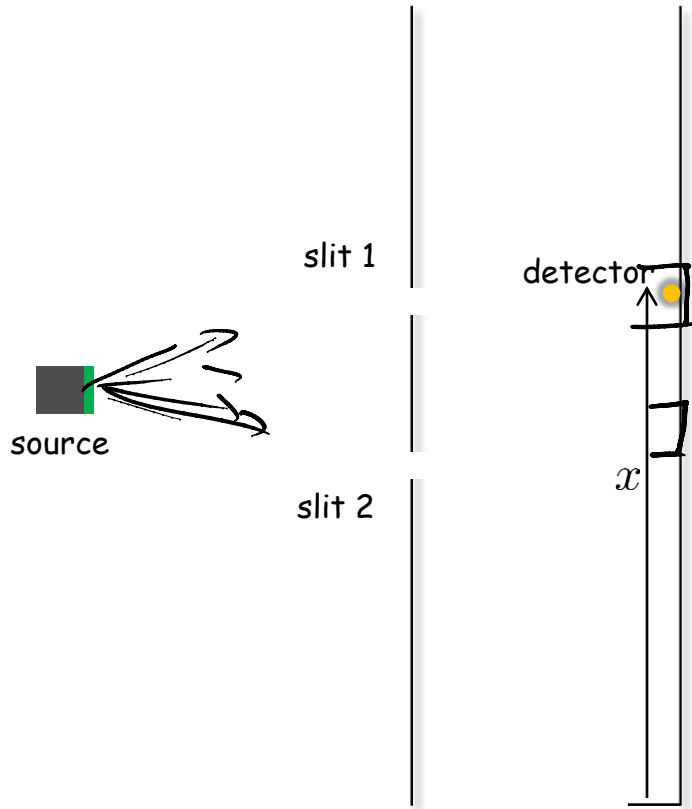


electrons / photons

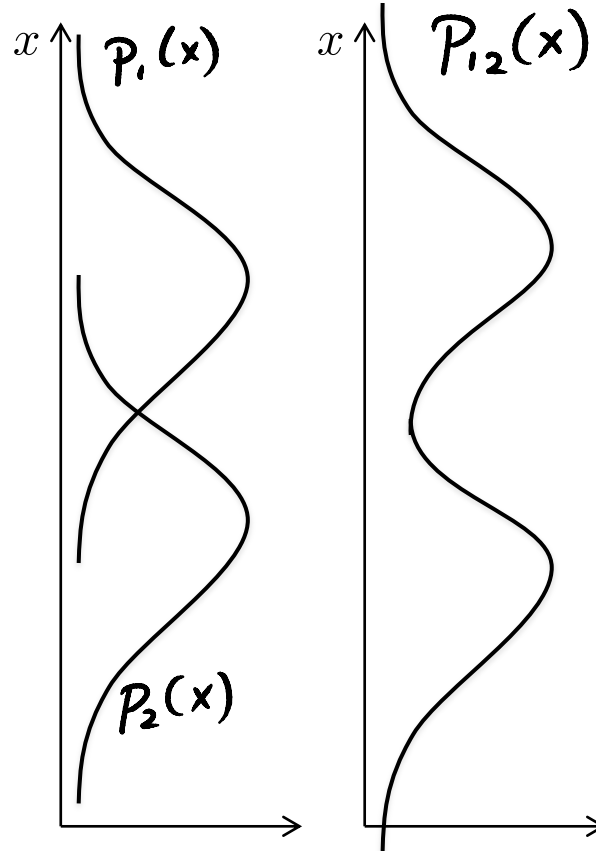
bullets

water waves

Double-slit experiment

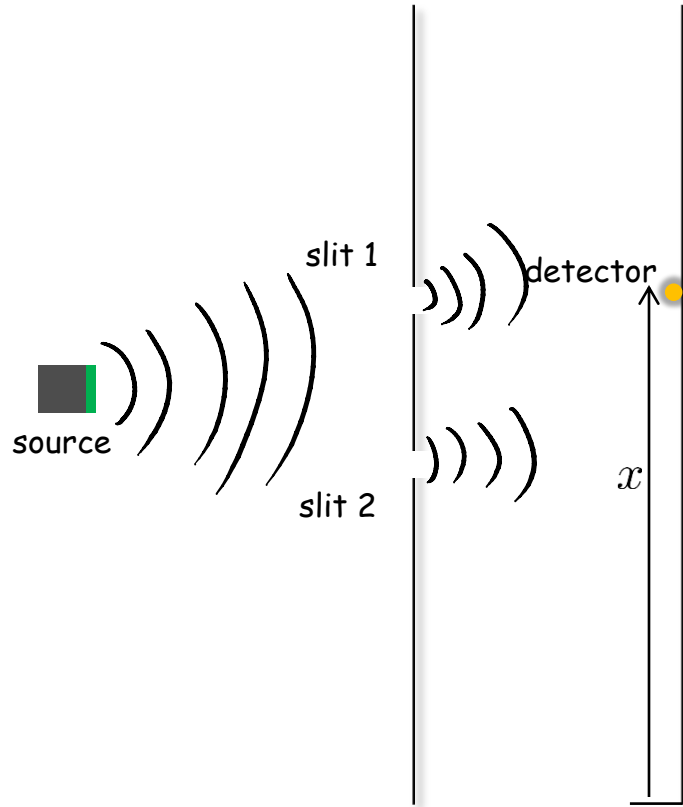


Bullets

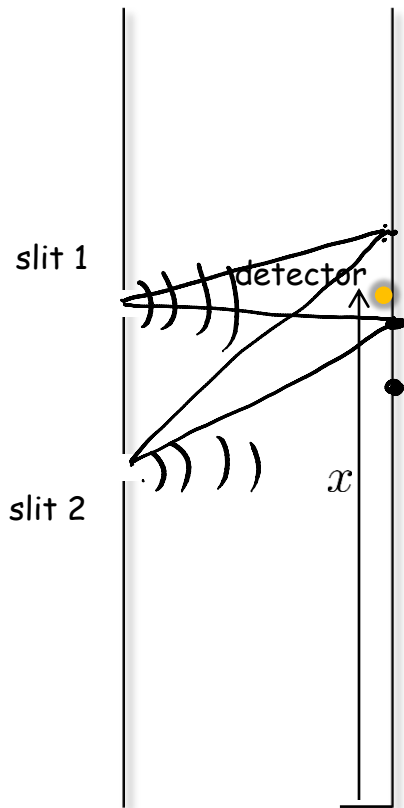


Double-slit experiment

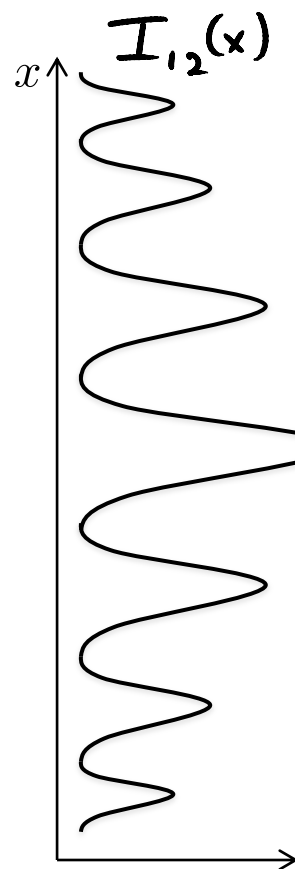
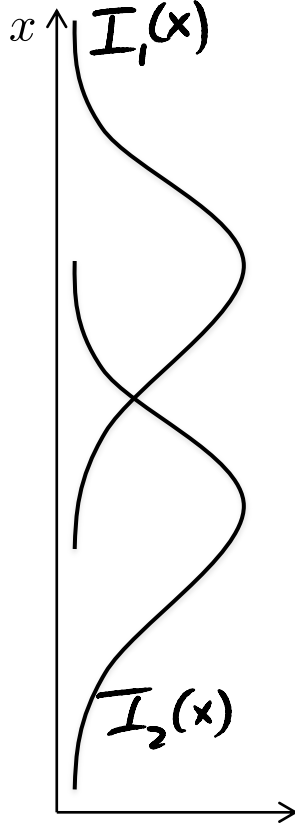
Waves



Double-slit experiment



Waves



$$I_{12}(x) \neq I_1(x) + I_2(x)$$

$$I(x) = h(x)^2$$

$$h_{12}(x) = h_1(x) + h_2(x)$$

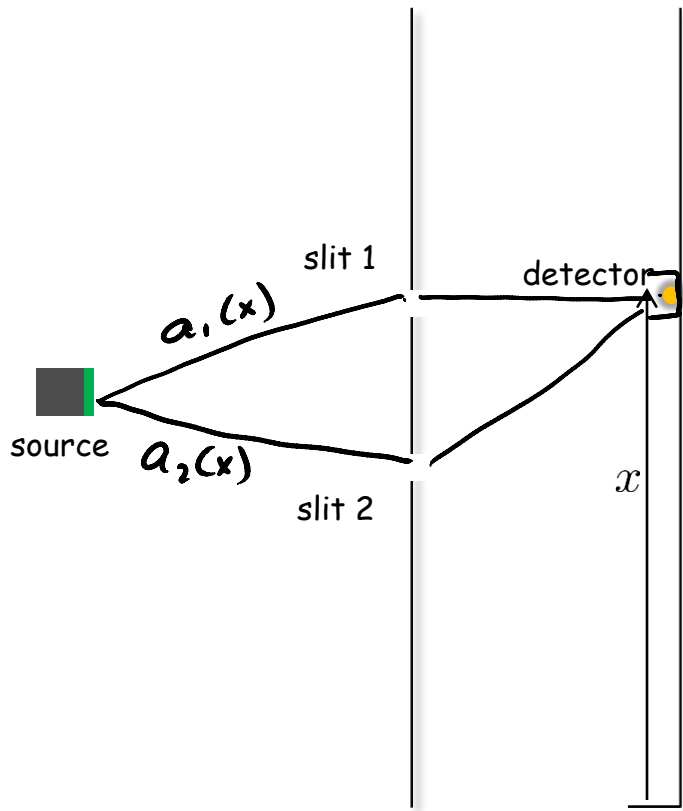
$$I_{12}(x) = h_{12}(x)^2$$

$$\neq h_1(x)^2 + h_2(x)^2$$

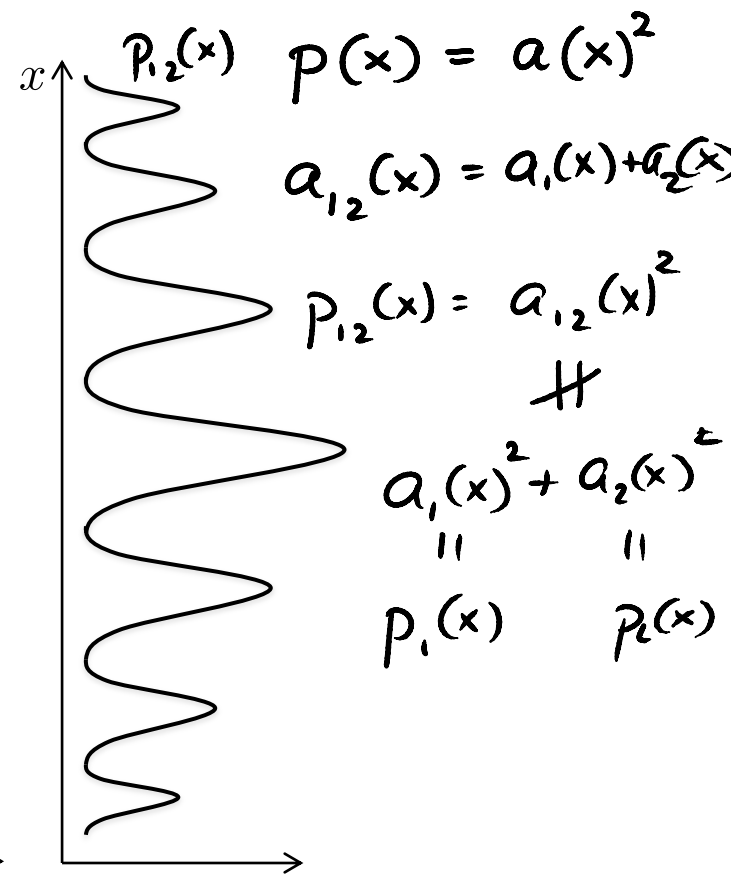
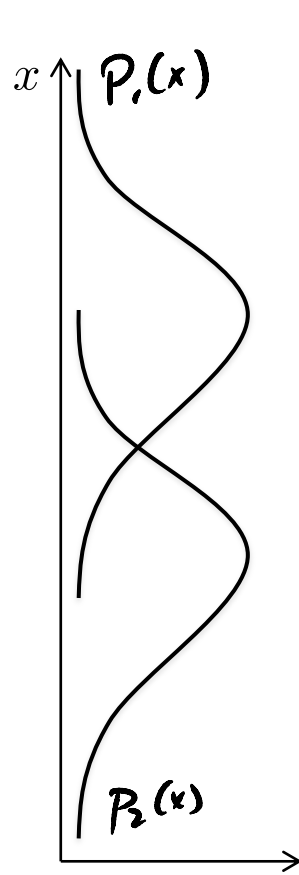
$$\parallel I_1(x)$$

$$\parallel I_2(x)$$

Double-slit experiment



Electrons/Photons $P_{1,2}(x) \neq P_1(x) + P_2(x)$



- If we want to describe what happens in an atomic event, we have to realize that the word "happens" can only apply to the observation, not to the state of affairs between two observations.

Heisenberg (1958)

- Do not keep saying to yourself, if you can possibly avoid it, 'But how can it possibly be like that?' because you will go down the drain into a blind alley from which nobody has yet escaped. Nobody knows how it can be like that.

Feynman (1965)

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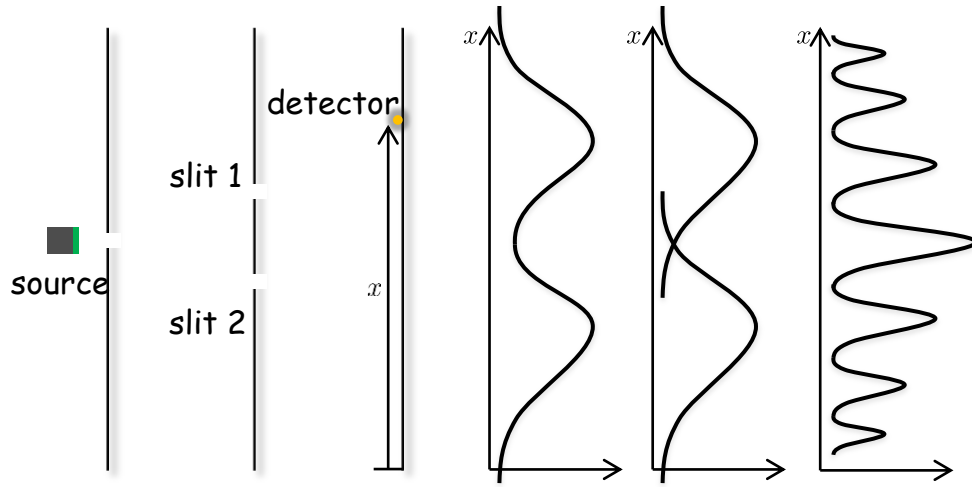
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$$\frac{1}{\sqrt{2}} |\text{cat}\rangle + \frac{1}{\sqrt{2}} |\text{dog}\rangle$$

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Double-slit experiment



$$I_1(x) = h_1(x)^2$$

$$I_2(x) = h_2(x)^2$$

$$h_{1,2}(x) = h_1(x) + h_2(x)$$

$$I_{1,2} = h_{1,2}^2 = (h_1(x) + h_2(x))^2$$

$$P_1(x) = |a_1(x)|^2$$

$$P_2(x) = |a_2(x)|^2$$

$$a_{1,2}(x) = a_1(x) + a_2(x)$$

Bullets

Discrete

Prob arrival

$$P_{1,2} = P_1 + P_2$$

No interference

Water Waves

Continuous

Intensity

$$I_{1,2} \neq I_1 + I_2$$

Interference.

Photons/Electrons.

Discrete

Prob of arrival.

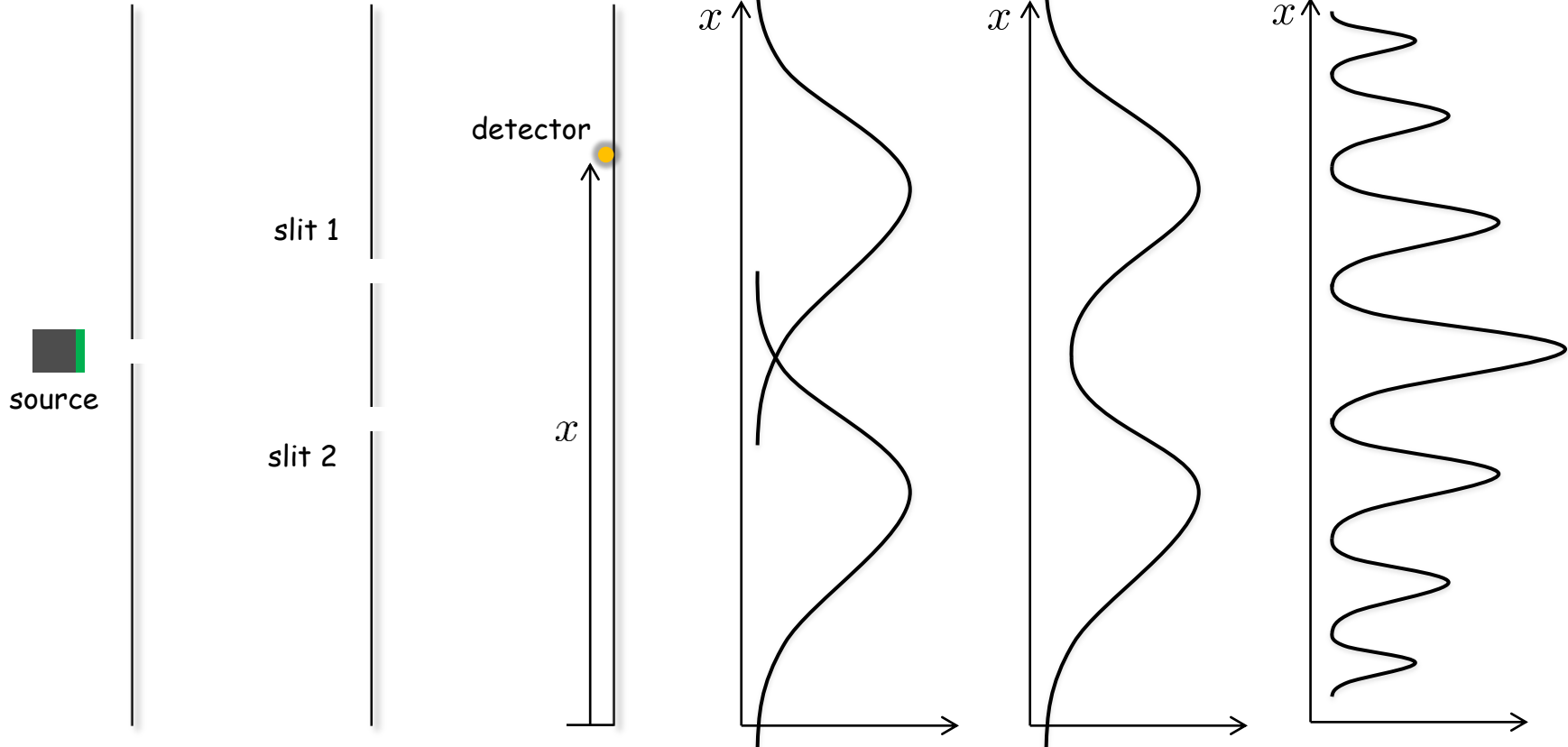
$$P_{1,2} \neq P_1 + P_2$$

Interference.

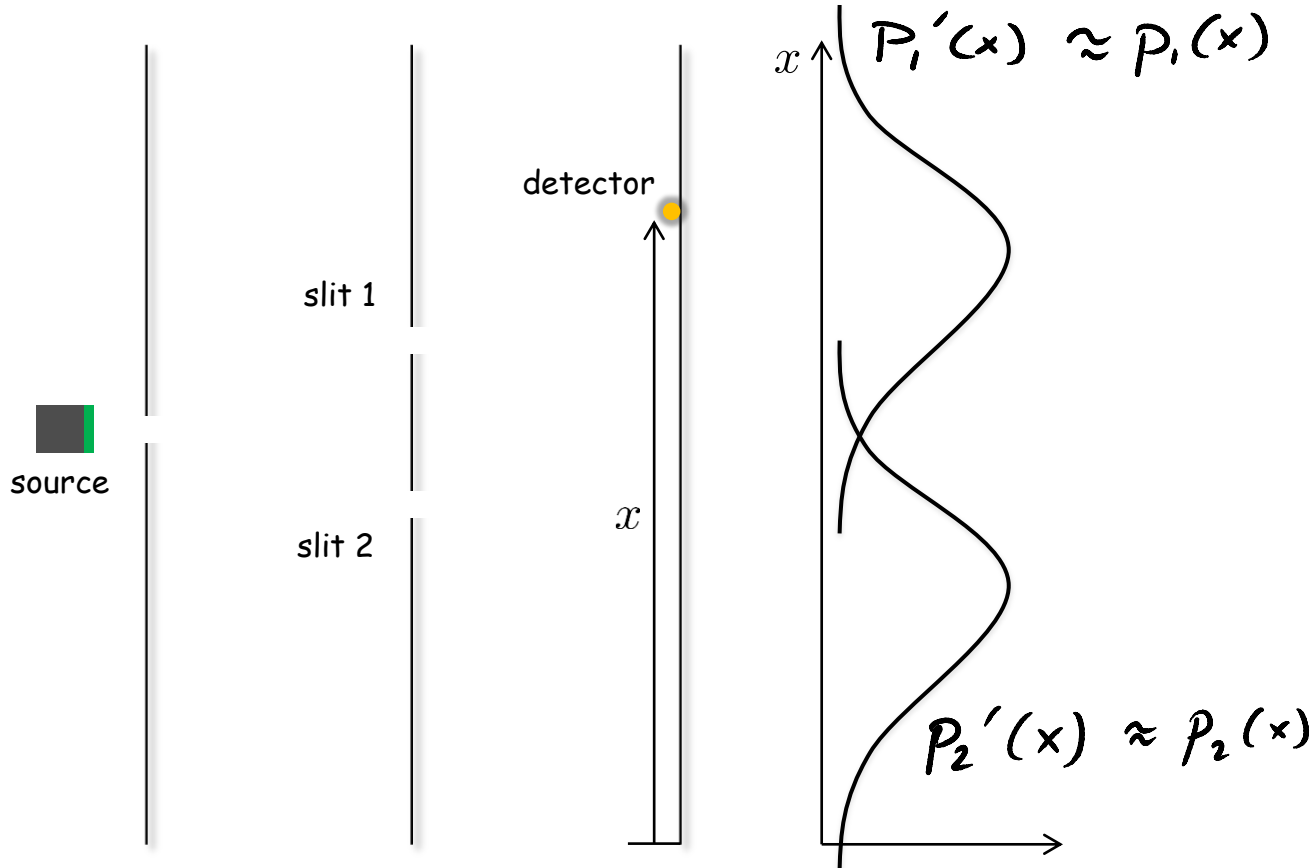
Proposition: The electron went through slit 1 or
it went through slit 2.

Proposition: The electron went through slit 1 or it went through slit 2.

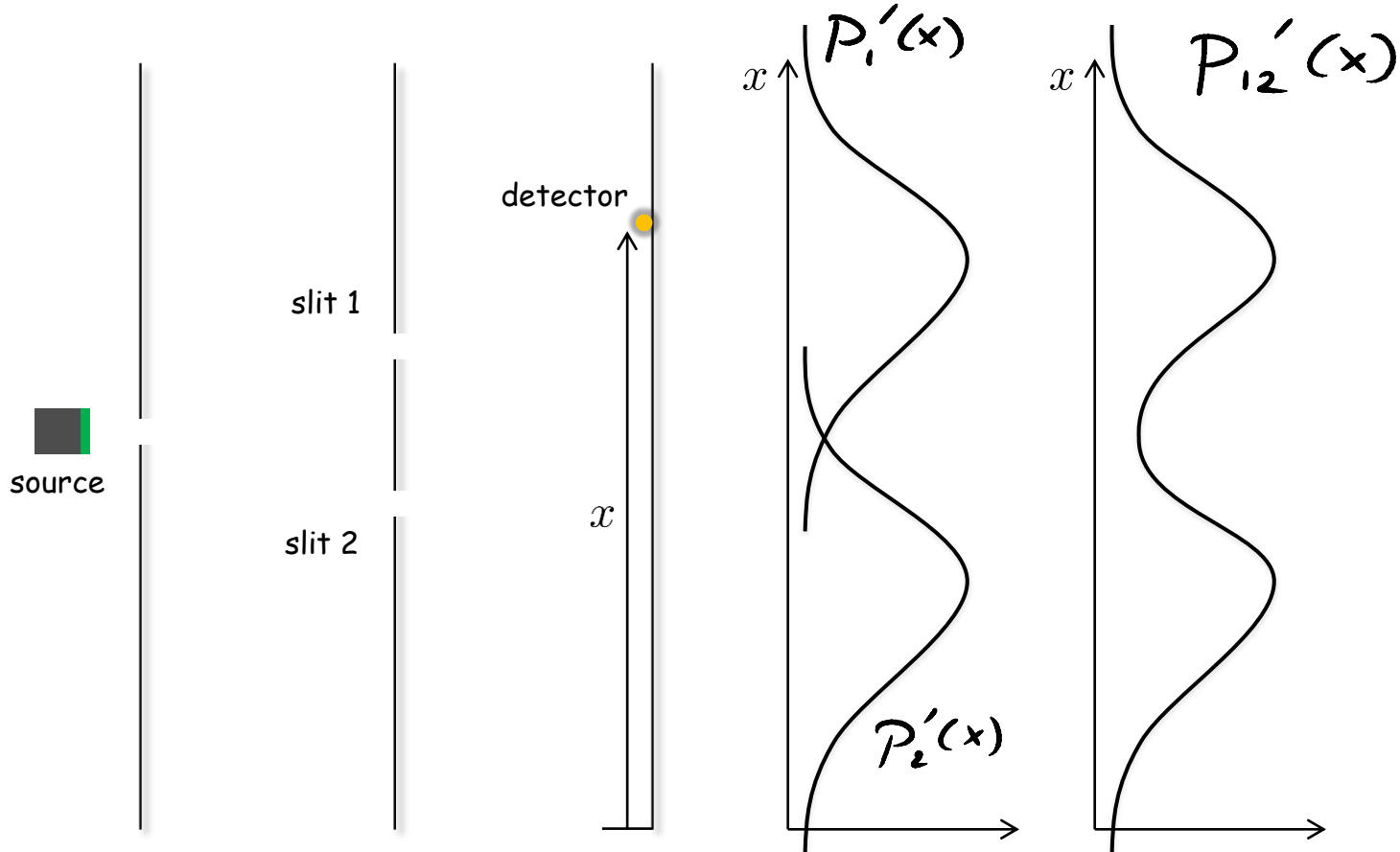
Test



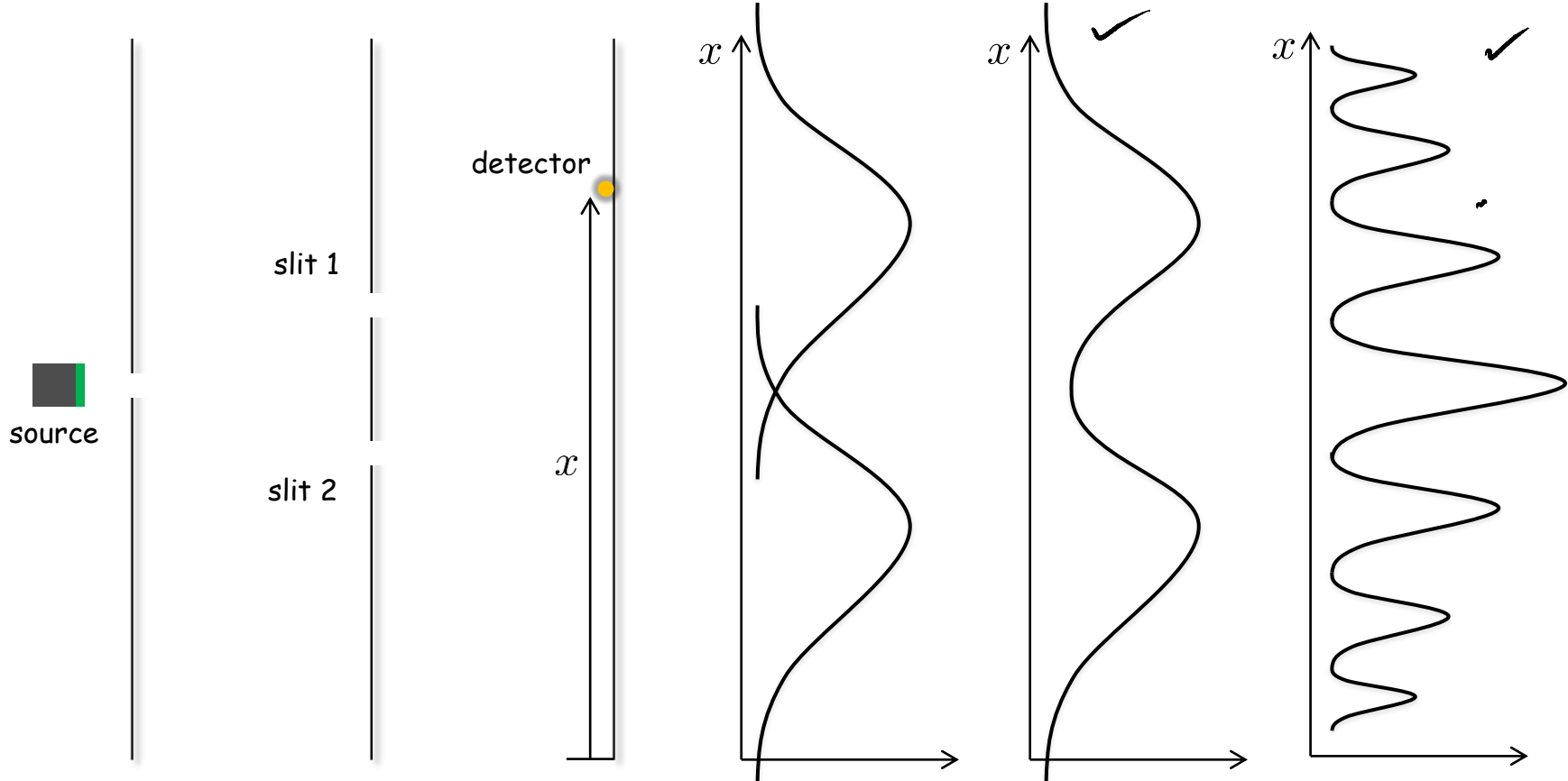
Proposition: The electron went through slit 1 or it went through slit 2.



Proposition: The electron went through slit 1 or it went through slit 2.



Proposition: The electron went through slit 1 or it went through slit 2.



Heisenberg's Uncertainty Principle

- An electron is delicate.
- Measurement disturbs system.
- Make light fainter:
 - But light is quantized
- Heisenberg's Uncertainty Principle: impossible to design apparatus that detects which slit without disturbing interference pattern.

Probability inherent in QM: suppose you could predict which slit the electron will go through:

